

CHM 421/621

Statistical Mechanics

Lecture 3

Introduction and Review

Lecture Plan

An example application of Statistical Mechanics (ctd.):

Calculation of average energy

A familiar example

Classical microstates - set of instantaneous coordinates and velocities.

System: Consider a system of N particles moving independently of each other and inside a closed container of volume V . The container (and hence the particles) is assumed to be in thermal contact with a reservoir at temperature T .

Model:

$$H(\{\mathbf{r}_i(t), \mathbf{p}_i(t)\}) = \sum_{i=1}^N \frac{1}{2m} p_i^2$$

\mathbf{p}_i Momentum of i^{th} particle

\mathbf{r}_i Position of i^{th} particle

Probability distribution:

$$\rho(\{\mathbf{r}_i, \mathbf{p}_i\}) = \frac{1}{Q(N, V, T)} \exp(-\beta H(\{\mathbf{r}_i, \mathbf{p}_i\}))$$

$$\beta = \frac{1}{k_B T}$$

A familiar example

Probability distribution:

$$\rho(\{\mathbf{r}_i, \mathbf{p}_i\}) = \frac{1}{Q(N, V, T)} \exp(-\beta H(\{\mathbf{r}_i, \mathbf{p}_i\}))$$

Meaning:

The probability of finding the system in a microstate with the momenta of the particles between \mathbf{p}_i and $\mathbf{p}_i + d\mathbf{p}_i$ and their positions between \mathbf{r}_i and $\mathbf{r}_i + d\mathbf{r}_i$ is given by

$$\rho(\{\mathbf{r}_i, \mathbf{p}_i\}) d^{3N} r_i d^{3N} p_i$$
$$\Rightarrow \int \rho(\{\mathbf{r}_i, \mathbf{p}_i\}) d^{3N} r_i d^{3N} p_i = 1$$

So what is $Q(N, V, T)$?

A familiar example

$$Q(N, V, T) = AV^N \left(\frac{2\pi m}{\beta} \right)^{\frac{3N}{2}}$$

Here A is a constant to ensure the correct dimensions. We will show later that

$$A = \frac{1}{h^{3N}}$$

$$Q(N, V, T) = \left(\frac{V}{\Lambda^3} \right)^N$$

With $\Lambda = \frac{h}{\sqrt{2\pi m k_B T}}$

In this case, the normalisation equation becomes

$$\frac{1}{h^{3N}} \int \rho(\{\mathbf{r}_i, \mathbf{p}_i\}) d^{3N} r_i d^{3N} p_i = 1$$

Average Energy

Calculate

$$U = \langle H \rangle = \frac{1}{h^{3N}} \int d\mathbf{r} \int d\mathbf{p} H(\{\mathbf{r}_i, \mathbf{p}_i\}) \rho(\{\mathbf{r}_i, \mathbf{p}_i\})$$

Answer

$$U = \frac{3}{2} N k_B T$$

Average Energy

$$\begin{aligned} U = \langle H \rangle &= \frac{1}{h^{3N}} \int d\mathbf{r} \int d\mathbf{p} H(\{\mathbf{r}_i, \mathbf{p}_i\}) \rho(\{\mathbf{r}_i, \mathbf{p}_i\}) \\ &= \frac{1}{Qh^{3N}} \int d\mathbf{r} \int d\mathbf{p} \left(\sum_{i=1}^N \frac{p_i^2}{2m} \right) \exp \left(-\beta \sum_{i=1}^N \frac{p_i^2}{2m} \right) \\ &= \frac{1}{Qh^{3N}} \left(\prod_{i=1}^N \int_V d\mathbf{r}_i \right) \int d\mathbf{p} \left(\sum_{i=1}^N \frac{p_i^2}{2m} \right) \exp \left(-\beta \sum_{i=1}^N \frac{p_i^2}{2m} \right) \\ &= \frac{1}{Qh^{3N}} V^N \int d\mathbf{p} \left(\sum_{i=1}^N \frac{p_i^2}{2m} \right) \exp \left(-\beta \sum_{i=1}^N \frac{p_i^2}{2m} \right) \end{aligned}$$

Average Energy

$$= \frac{1}{Qh^{3N}} V^N \sum_{j=1}^N \int d\mathbf{p} \frac{p_j^2}{2m} \exp \left(-\beta \sum_{i=1}^N \frac{p_i^2}{2m} \right)$$

$$= \frac{N}{Qh^{3N}} V^N \int d\mathbf{p} \frac{p_1^2}{2m} \exp \left(-\beta \sum_{i=1}^N \frac{p_i^2}{2m} \right)$$

$$= \frac{N}{Qh^{3N}} V^N \left(\int \prod_{i=1}^N d\mathbf{p}_i \frac{p_1^2}{2m} \exp \left(-\beta \frac{p_i^2}{2m} \right) \right)$$

Since all p_j are being integrated over each integral in the sum is identical

Average Energy

$$= \frac{N}{Qh^{3N}} V^N \prod_{i=2}^N \left(\int d\mathbf{p}_i \exp \left(-\beta \frac{p_i^2}{2m} \right) \right) \int d\mathbf{p}_1 \frac{p_1^2}{2m} \exp \left(-\beta \frac{p_1^2}{2m} \right)$$

$$= \frac{N}{Qh^{3N}} V^N \left(\frac{2\pi m}{\beta} \right)^{\frac{3(N-1)}{2}} (2m)^{\frac{3}{2}} \int d\mathbf{y} y^2 \exp(-\beta y^2)$$

where we have substituted $\mathbf{y} = \frac{\mathbf{p}}{\sqrt{2m}}$

Average Energy

Now

$$\begin{aligned} & \int d\mathbf{y} \, y^2 \exp(-\beta y^2) \\ &= \sum_{\nu=1}^3 \left(\prod_{\mu \neq \nu} \int dy_{\mu} \exp(-\beta y_{\mu}^2) \right) \int dy_{\nu} \, y_{\nu}^2 \exp(-\beta y_{\nu}^2) \\ &= \sum_{\nu=1}^3 \frac{\pi}{\beta} \int dy_{\nu} \, y_{\nu}^2 \exp(-\beta y_{\nu}^2) \\ &= \frac{3\pi}{\beta} \int dy_1 \, y_1^2 \exp(-\beta y_1^2) = \frac{3}{2\beta} \left(\frac{\pi}{\beta} \right)^{\frac{3}{2}} \end{aligned}$$

Using this formula in the previous equations and setting $Q = \left(\frac{V}{\Lambda^3} \right)^N$ we get ...

Average Energy

$$U = N \Lambda^{3N} \left(\frac{2\pi m}{\beta \hbar^2} \right)^{\frac{3N}{2}} \frac{3}{2\beta}$$

$$= \boxed{\frac{3}{2} N k_B T}$$

Using

$$\Lambda = \frac{h}{\sqrt{2\pi m k_B T}}$$

$$\beta = \frac{1}{k_B T}$$

Which is the desired result!