State of a quantum system - Postulate 1

Lecture 2

CHM 439

Quantum Chemistry

Varadharajan Srinivasan Dept. Of Chemistry IISER Bhopal

Lecture Plan

- About waves and their wave function (Courtesy: Prof. M. Weichman)
- Back to the double-slit experiment
- Matter waves and their wave function
- The first postulate of QM

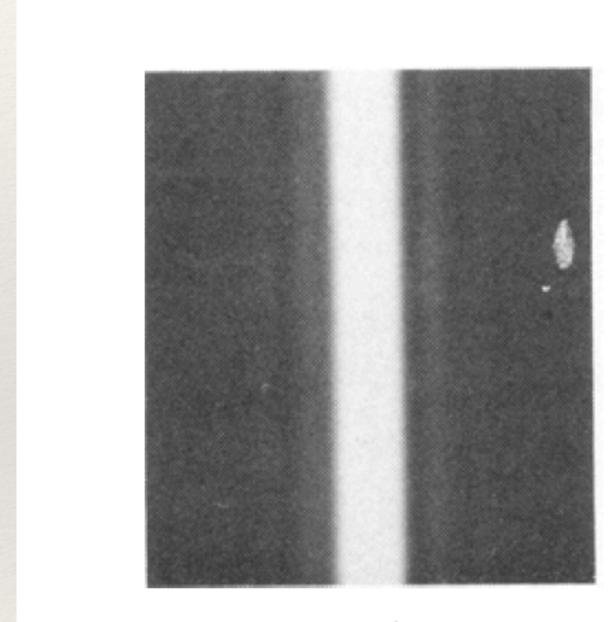
Summary of the double-slit experiment with electrons

- Interference pattern observed at the detector when both slits are open.
- Closing either slit results in "particle-like" distribution at the detector (predominantly one bump).*
- Slowing down the rate of electrons (one electron at a time) still gives the interference pattern.
- If we try to observe which slit the particle goes through we get back the "particle-like" distribution.
- The fate of each electron cannot be predicted but the final interference patten is exactly reproducible (for the same source and same slits).

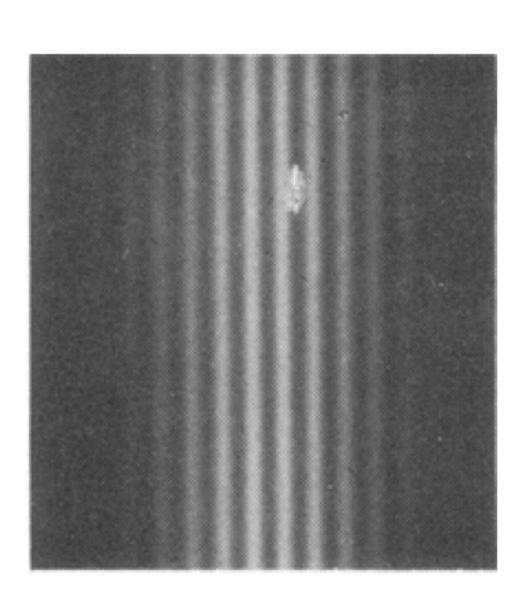
N.B. Here, "particle-like" actually refers to a single-slit interference pattern!

Interference of matter waves for larger particles

Electron diffraction through multiple slits (Am. J. Phys. 42, 4, 1974)



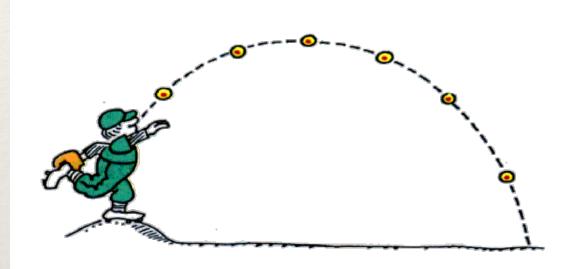
Single slit

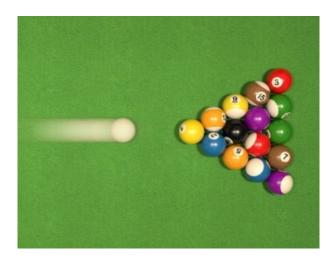


Two slits

Classical mechanics

particles





Newton's laws $F = ma = m\ddot{x}$

$$F = ma = m\ddot{x}$$

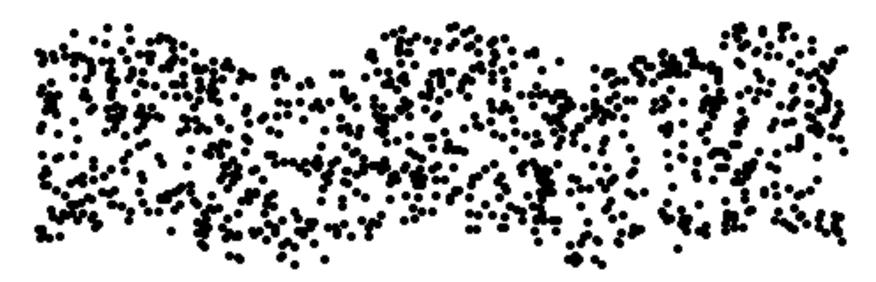
can calculate trajectories in time:

$$x(t), \dot{x}(t), \ddot{x}(t)$$

Courtesy: Prof. M. Weichman

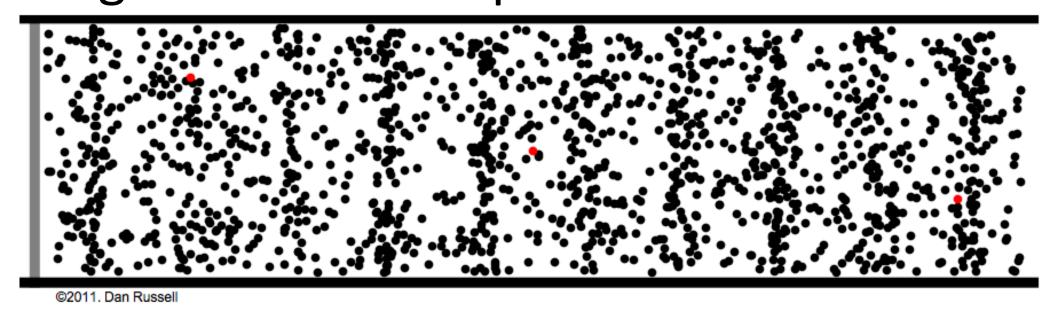
waves

transverse waves





longitudinal or compression waves

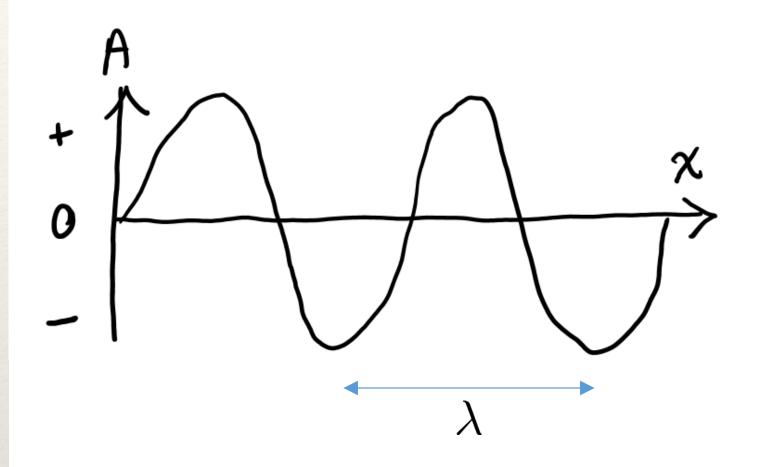




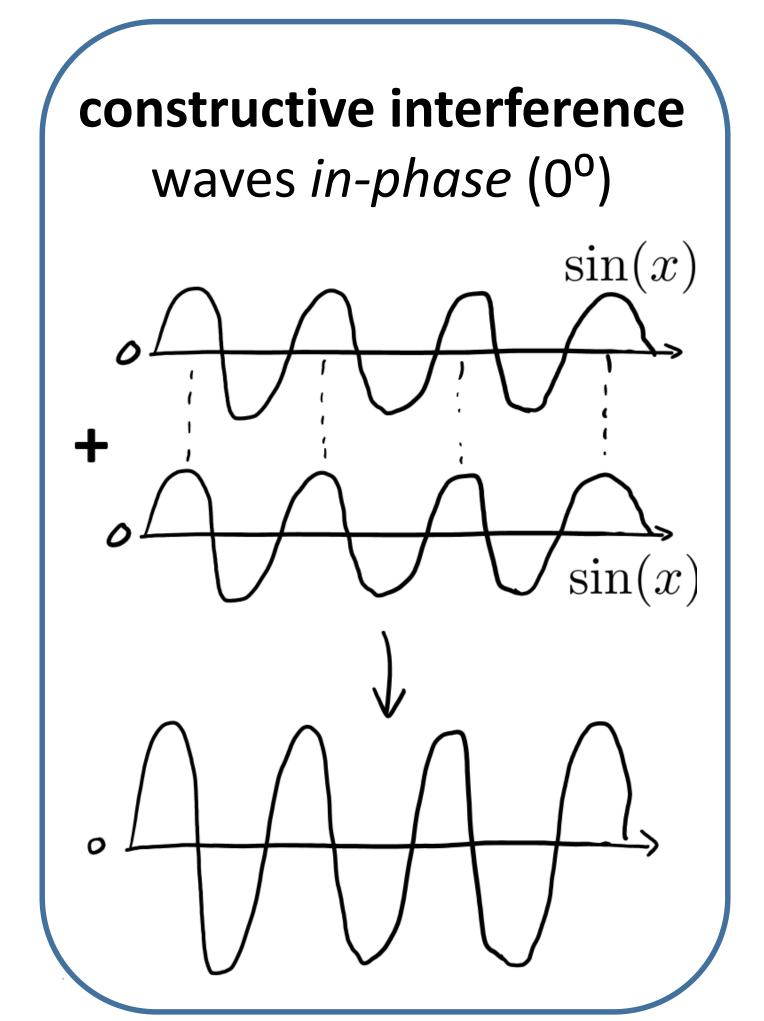
can calculate displacement as a function of time:

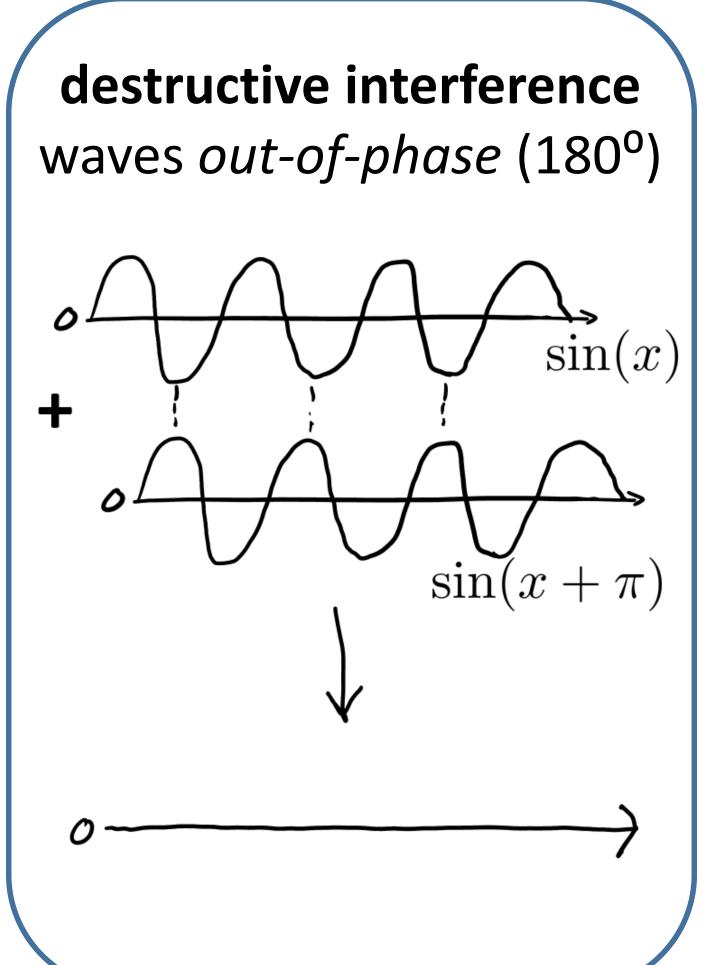
$$\frac{\partial^2 g}{\partial x^2} = \frac{1}{v^2} \frac{\partial^2 g}{\partial t^2} \quad \to \quad g(x, t)$$

Classical wave phenomena



- displacement or oscillation about zero mean
- amplitude has a sign
- interference, diffraction effects depend on relative *phases* of waves



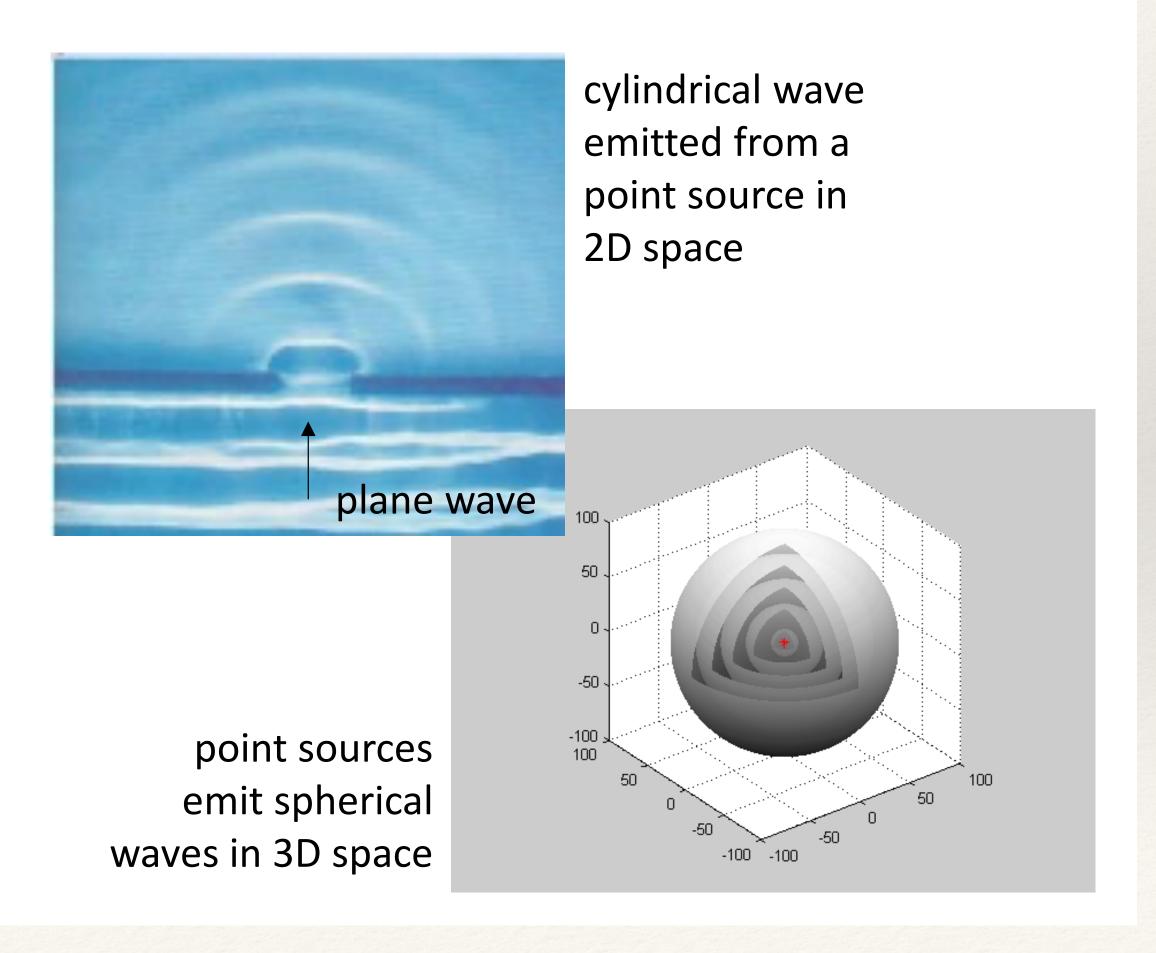


Classical wave phenomena

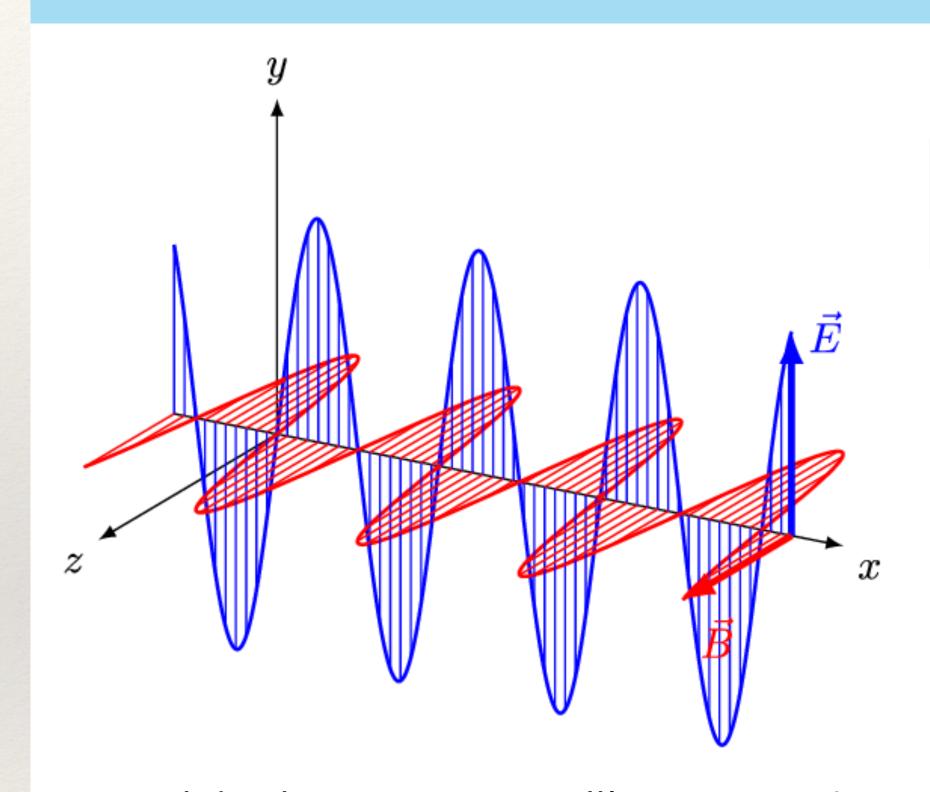
waves are delocalized



wave fronts can be curved



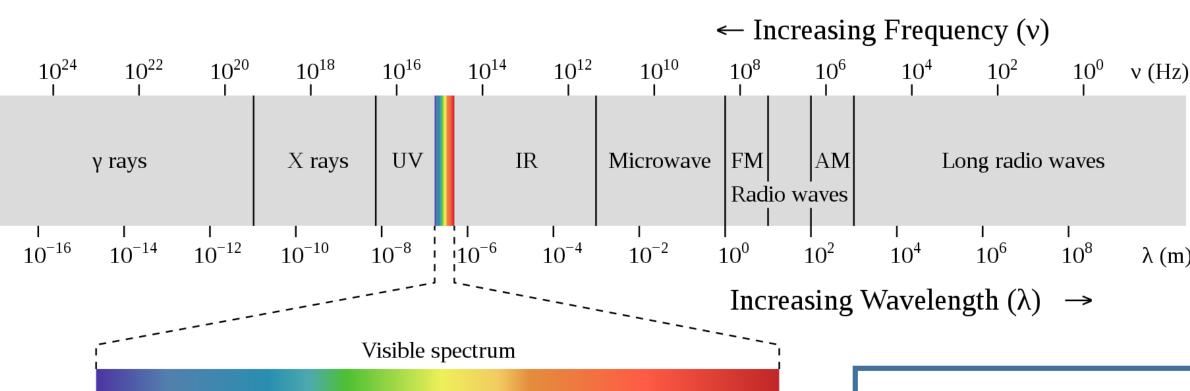
Light is an electromagnetic wave

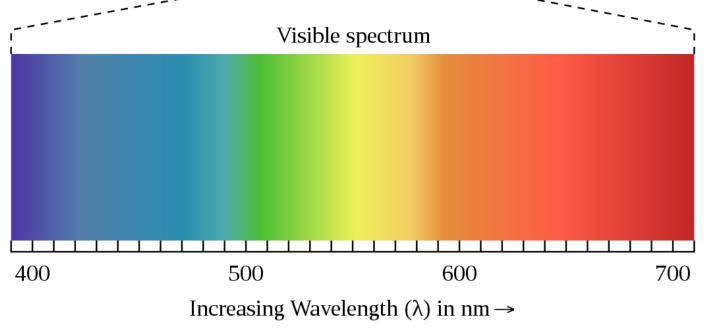


a valid solution to Maxwell's equations!

$$E_y(x,t) \sim \sin(kx - \omega t)$$

 $B_z(x,t) \sim \sin(kx - \omega t)$





wavelength,
$$\lambda$$

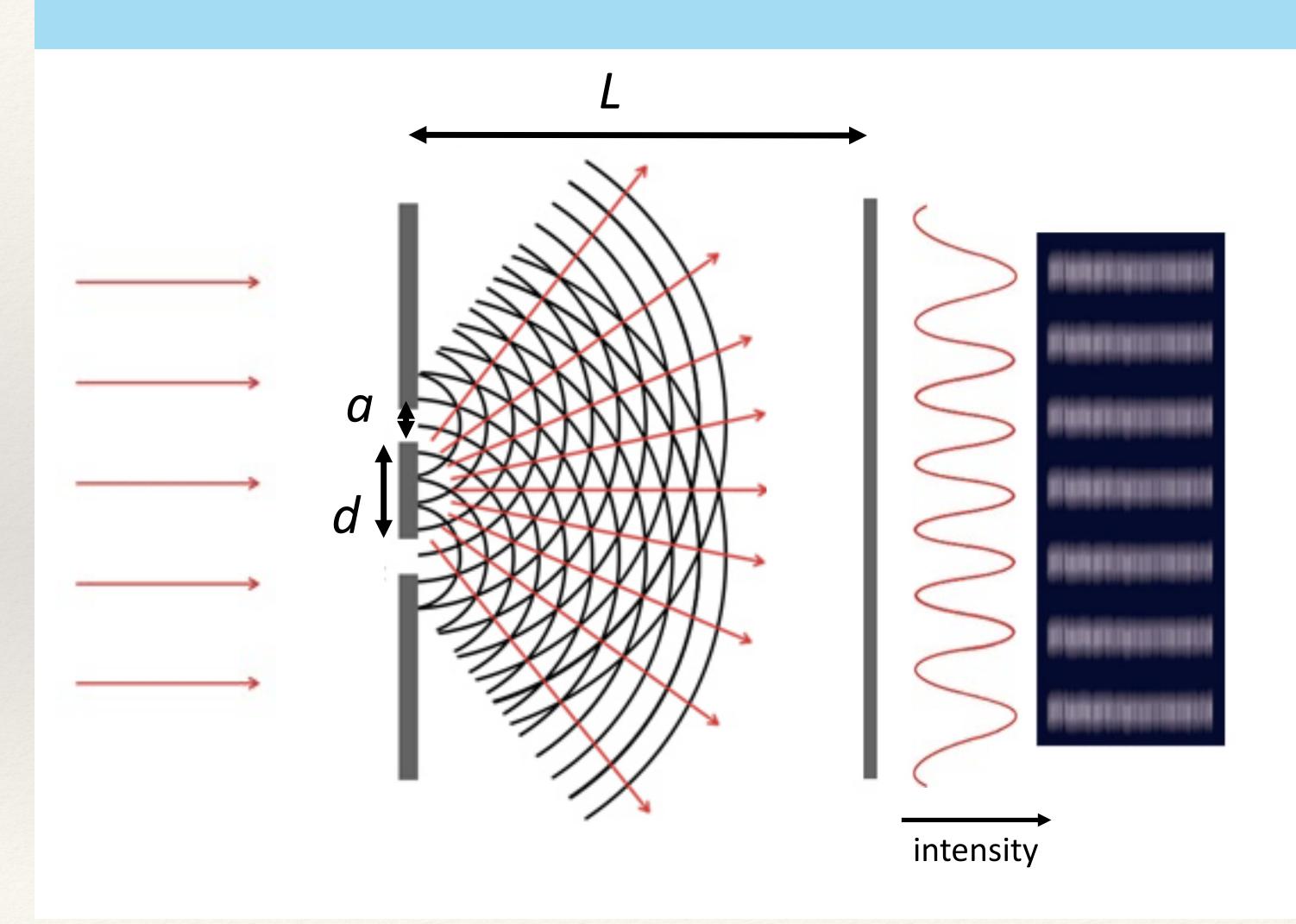
frequency,
$$\nu = \frac{c}{\lambda}$$

wavenumber,
$$\tilde{v} = \frac{1}{\lambda}$$

$$E = h\nu$$

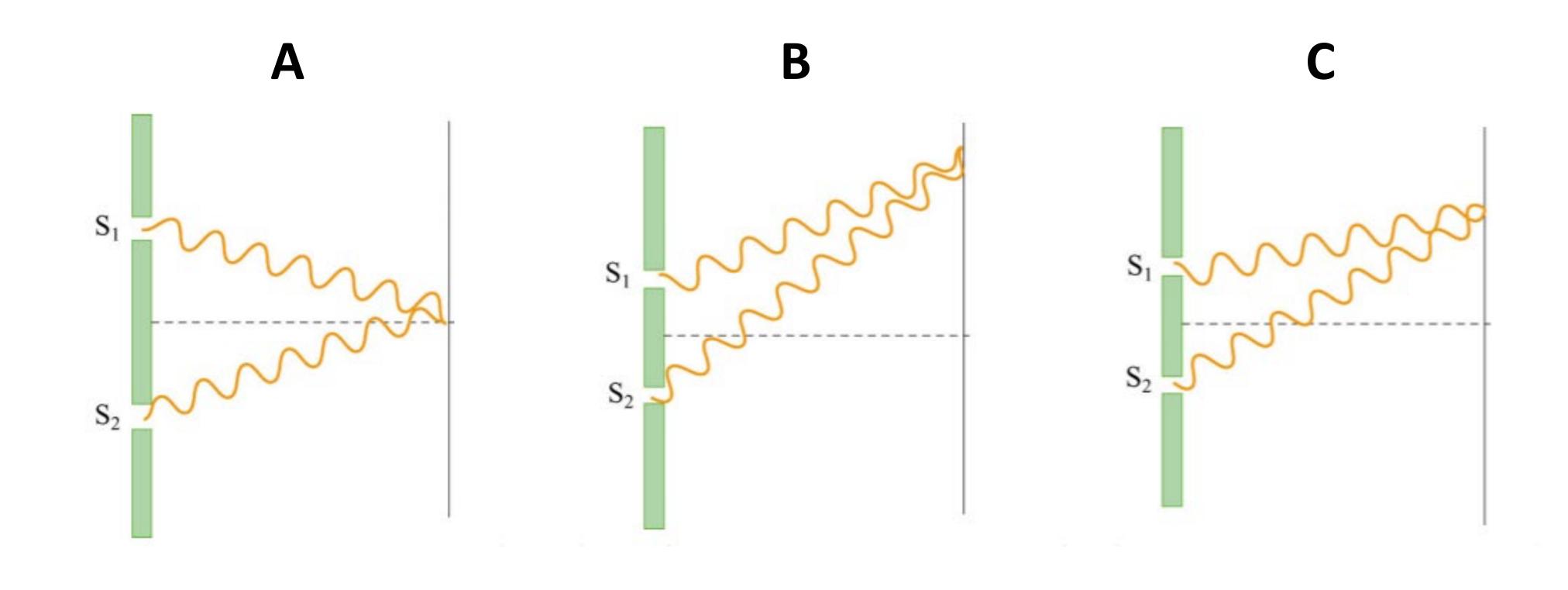
$$h = 6.626 \times 10^{-34} \,\mathrm{J \cdot s}$$

Classical diffraction of light through a double slit

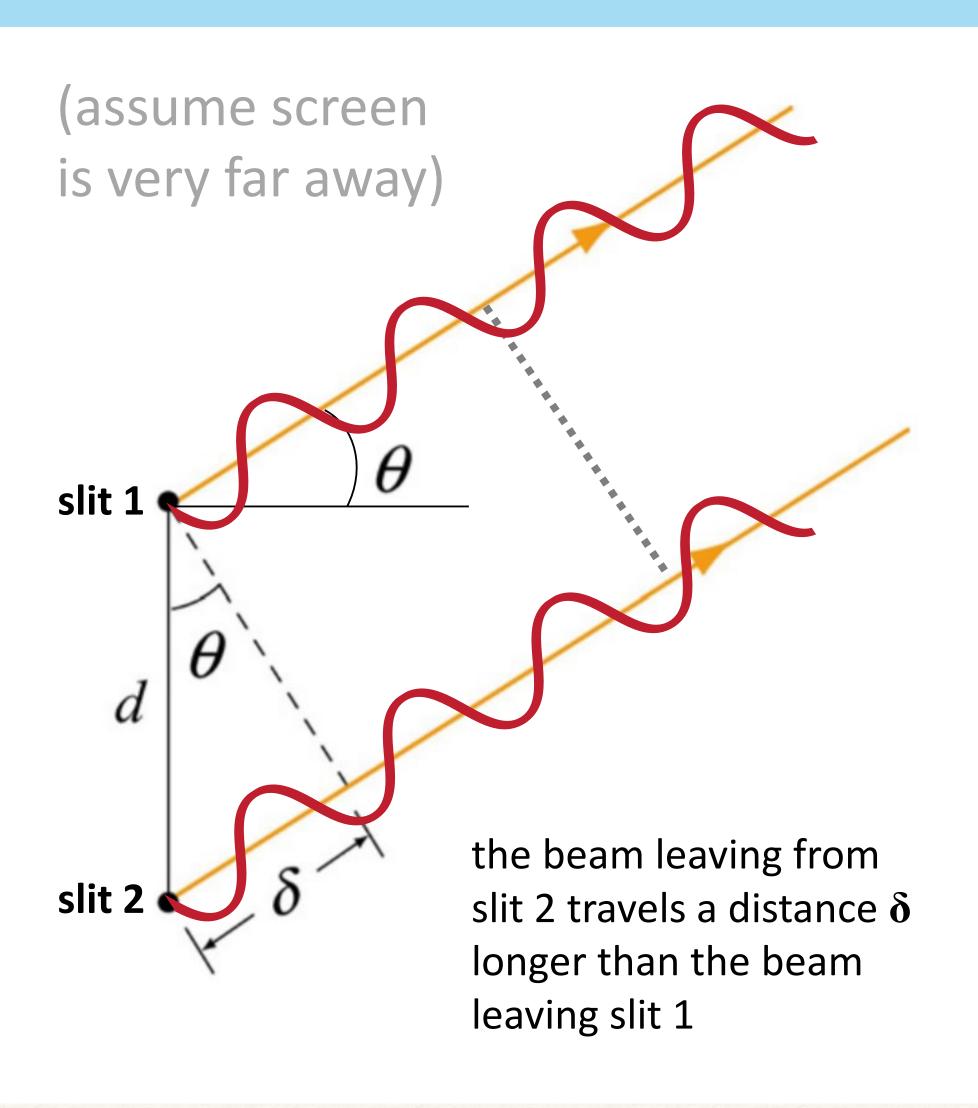


- each slit becomes a new point source of light
- two sources interfere with one another
- destructive interference at minima, constructive at maxima

Which positions on the screen are bright or dark?



Determining the fringe positions mathematically



waves interfere constructively when difference between the pathlengths traveled from each slit to the screen is a round integer multiple of the wavelength

$$\delta = d\sin\theta = m\lambda, \quad m = 0, \pm 1, \pm 2, \dots$$

$$\sin heta = m \lambda/d$$
 \longrightarrow bright spots appear at angles: $heta_{
m bright} = \arcsin(m \lambda/d)$

$$\theta_{\text{bright}} = \arcsin(m\lambda/d)$$

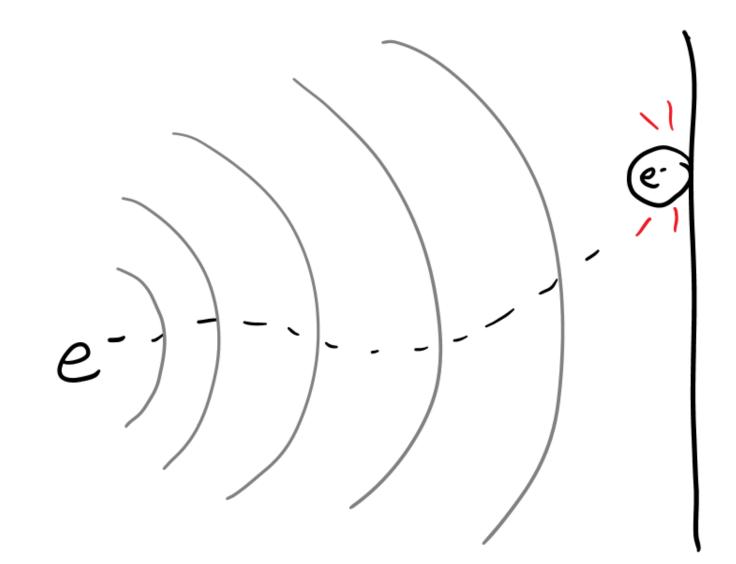
how might we alter this expression to locate the angles where dark spots appear?

$$d \sin \theta = (\underline{m+1/2})\lambda, \quad m = 0, \pm 1, \pm 2, \dots$$

$$\theta_{\rm dark} = \arcsin((m+1/2)\lambda/d)$$

Quantum mechanical wave-particle duality

q small particle (e⁻, atom, molecule) shows properties of both waves and particles:

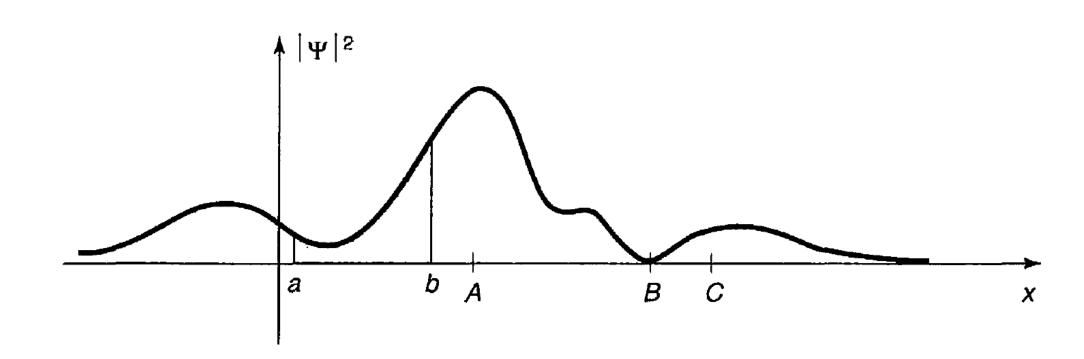


imagine if sound waves behaved like this!

quantum particles are governed by a *wavefunction*

 describes the probability distribution of where we expect to find the particle

$$\psi(x,t)$$



$$\int_a^b |\psi(x,t)|^2 dx = \begin{cases} \text{probability of finding the particle between } a \\ \text{and } b \text{, at time } t \end{cases}$$

Summary of the double-slit experiment with electrons

Non-classical result $(P_{12} /= P_1 + P_2)$

- Interference pattern observed at the detector when both slits are open.
- Closing either slit results in "particle-like" distribution at the detector (predominantly one bump).*
- Slowing down the rate of electrons (one electron at a time) still gives the interference pattern. P_{12} results from individual electron events!
- If we try to observe which slit the particle goes through we get back the "particle-like" distribution.

 Emergence of an uncertainty principle.
- The fate of each electron cannot be predicted but the final interference patten is exactly reproducible (for the same source and same slits). A new definition for the state of a system.

N.B. Here, "particle-like" actually refers to a single-slit interference pattern!

Acknowledgements

Wikipedia commons
Google images
YouTube
The textbooks of the course.
https://weichman.princeton.edu